

---

---

## ATMOSPHERIC RADIATION, OPTICAL WEATHER, AND CLIMATE

---

---

# The Scale of Ozone Destruction in Clouds

M. Yu. Arshinov<sup>a</sup>, B. D. Belan<sup>a, b</sup>, G. N. Tolmachev<sup>a</sup>, and A. V. Fofonov<sup>a</sup>

<sup>a</sup> Zuev Institute of Atmospheric Optics, Siberian Branch, Russian Academy of Sciences,  
pl. Akademika Zueva 1, Tomsk, 634021 Russia

<sup>b</sup> Tomsk State University, pr. Lenina 36, Tomsk, 634050 Russia

Received June 3, 2009

**Abstract**—Changes in the concentration of tropospheric ozone in clouds were investigated based on aircraft sensing data. Three ozonometers were used for the measurements—one chemiluminescent 3-02P and two UV 49C (Thermo Environment Inc., United States). The following types of clouds were studied: *Cu*, *Cu med.*, *St*, *Sc*, *As*, and *Ac*. The thickness of the cloud layers was 1.5 km on average and varied from 0.4 to 4.5 km. The ozone destruction in clouds was 11–15 ppb on average and ranged from 3 to 34 ppb; it changed nearly twofold depending on the cloud type. The estimation of the annual runoff of ozone in clouds has shown that it is close to the annual ozone balance in the troposphere.

DOI: 10.1134/S1024856010020041

## INTRODUCTION

Tropospheric ozone interacts directly with life forms, revealing its toxicant properties especially in the surface air layer. In large concentrations, it strongly inhibits the vital activity of plants and affects the human body in many ways. According to biological and medical investigations [1], ozone in the troposphere is a virulent poison; in addition to systemic toxicity, it exhibits such properties as mutagenicity, oncogenicity, and a radiomimetic effect (an effect on blood similar to ionizing radiation). Ozone exceeds such a well-known poison as hydrocyanic acid in its toxicity level. In addition to the impact on people and plants, ozone is an all-powerful oxidizer that damages rubber and oxidizes many metals, even of the platinum family [2].

Having a long lifetime in the atmosphere (from several days to several months) and strong absorption lines, tropospheric ozone can play a significant role in the greenhouse effect. According to [3], it contributes more than 8% to the total warming of the air caused by the absorption of solar radiation by greenhouse gases. Later estimates have shown that the contribution can be even larger.

Everything said requires a more detailed study of the balance of the ozone concentration in the troposphere. We considered the input part of the balance in previous works [4–7]. In addition, there is also the output part; together they form the ozone balance in the troposphere. Ozone runoff from the troposphere can result from the following processes: ozone photolysis, ozone interaction with gas components (ozonolysis), ozone interaction with aerosols, ozone sedimentation to the underlying surface, and precipitation erosion. The interaction of ozone with cloud aerosol is the least understood. These questions were studied theo-

retically in [8–14]. The number of experimental works is far less [15–17]; therefore, the aim of this work is to estimate the ozone destruction in clouds from experimental data.

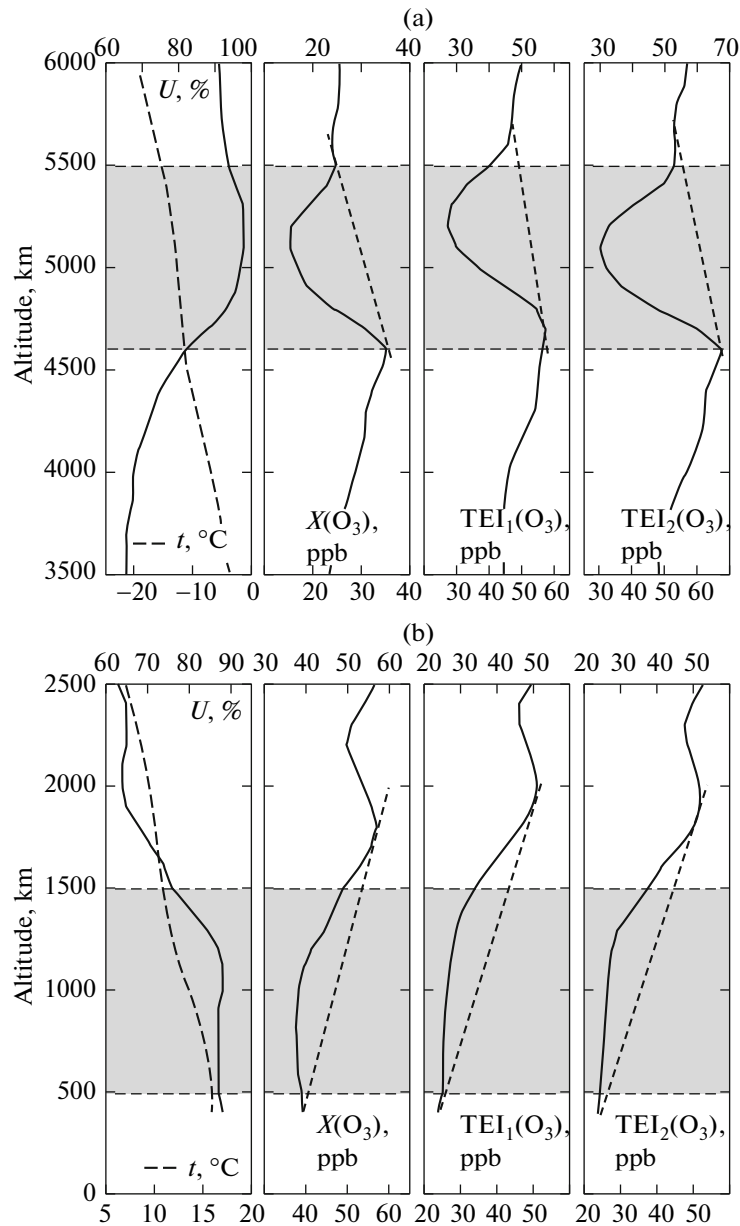
## 1. INSTRUMENTS AND TECHNIQUES

The measurements were carried out from onboard the “Optik-E” AN 30 aircraft during cloud crossing, climb, and descent.

The problem of ozone measurements in conditions deviating from the background ones was earlier discussed in [18–23]. In view of this, three onboard ozonometers operated simultaneously—one chemiluminescent 3-02P, designed at ZAO OPTEC (St. Petersburg, Russia), and two UV 49C (Thermo Environment Inc., United States). The ozonometers were calibrated using a GS-2 ozone generator (ZAO OPTEC). Air venting of the devices was carried out via Teflon tubes. Cloud crossing was controlled visually as well as by records of the meteoroparameters and aerosol number density.

Examples of the recording of the ozone concentration by the devices are shown in the figure (X corresponds to the chemiluminescent ozonometer, and TEI<sub>1</sub> and TEI<sub>2</sub> correspond to the UV ones). The distribution of the temperature and relative air humidity is shown in the left part of the figure; the cloud layer is grey.

As is seen from the Fig. 1, the ozone concentration either decreases inside the cloud layer or remains invariable up to the upper boundary. The concentration increases near the upper boundary. This seemingly occurs due to the involvement of air richer in ozone in the environmental exchange. The dashed lines show the deviation of the real distribution from



**Fig. 1.** Variations of the ozone concentration while crossing clouds: *As* near Khatanga on July 8, 2008 (a) and *Cu* near Yakutsk on July 23, 2008 (b).

the probable one in the case of the absence of the cloud layer.

The Fig. 1 also shows a good qualitative agreement between the readings of the three ozonometers. The quantitative criteria for 158 cases of cloud crossing by the aircraft are given in Table 1.

The single-shot accuracy of the 3-02P ozonometer is 15% and of the Thermo Environment Inc. (Model-49C) ozonometer is  $\pm 1$  ppb; in view of this, the mean deviations in Table 1 perfectly fall within the error range. In addition, the measurements can differ profoundly in certain cases. This especially concerns the X-TEI<sub>2</sub> pair, where the difference in the readings

reached +6 and  $-19$  ppb. We will not dwell on this fact, because it is not the subject of our study. Let us assume that the above three ozonometers satisfactorily reflect the ozone distribution in the cloud layers.

## 2. MEASUREMENT RESULTS AND DISCUSSION

First, consider the measurements of the ozone concentration in the cloud layer independently of the cloud type (Table 2).

As follows from Table 2, the measurements were carried out in clouds from 0.4 to 4.5 km in thickness at

**Table 1.** Deviations (ppb) between the readings of the ozonometers while crossing clouds

Deviation	X-TEI <sub>1</sub>	X-TEI <sub>2</sub>	TEI <sub>1</sub> -TEI <sub>2</sub>
Average	2.6	4.7	2.2
Rms	1.4	1.4	1.0
Minimum	-12	-19	-10
Maximum	6	6	1

**Table 2.** Variations of the ozone concentration while crossing clouds

Deviation	Variation of the ozone concentration, ppb			Cloud layer thickness, km
	Ozonometer			
	X	TEI <sub>1</sub>	TEI <sub>2</sub>	
Average	11	12	15	1.5
Rms	5	6	8	0.8
Minimum	3	3	6	0.4
Maximum	25	24	34	4.5

**Table 3.** Relative variation of the ozone concentration in clouds

Ozonometer	X	TEI <sub>1</sub>	TEI <sub>2</sub>
Relative variation, %	22	24	27

**Table 4.** Variations of the ozone concentration in clouds of different types

Deviation	Cloud type					
	<i>Cu</i>	<i>Cu med.</i>	<i>St</i>	<i>Sc</i>	<i>As</i>	<i>Ac</i>
Average ± rms, ppb	11 ± 5	14 ± 5	10 ± 5	9 ± 2	11 ± 5	8 ± 2
Relative, %	17	25	17	27	27	17
Range, ppb	7–16	8–25	4–15	7–11	3–18	5–10
The number of events	13	26	25	13	30	26

**Table 5.** Estimates of the power of the sources and ozone runoff in the troposphere (10<sup>12</sup> g/year) and budget (10<sup>12</sup> g)

Components	The Northern Hemisphere	The Southern Hemisphere	Earth
Inflow from the stratosphere	420	230	650
	430	220	650
Photochemical formation	770–2050	320–450	990–2500
	330–1240	180–690	310–1930
Photochemical runoff	1330	1110	2440
Runoff to the Earth's surface	940–1300	360–760	1300–2100
	650	430	1080
Budget	94	47	141
	158	83	241

an average value of 1.5 km. The average decrease in the ozone concentration in a cloud was 11, 12, and 15 ppb for each ozonometer, respectively; the coefficient of the variation reached 50%. The obtained values are significantly higher than in [15–17]. This can very well be due to the physical and geographical features of the region under study, as well as the experimental periods.

To partly exclude the climatic effect, we give the relative variations of the ozone concentration in clouds (Table 3). Here, the variation value is normalized to the average ozone concentration in the cloud marked by the dashed line in the Fig. 1.

It follows from Table 3 that the X ozonometer shows the minimum while the TEI<sub>2</sub> shows the maximum values; the relative ozone variation is 22–27% of its atmospheric concentration.

Variations of the ozone concentration in certain types of clouds can be inferred from Table 4 built on the basis of the 3-02P ozonometer data.

It is seen from Table 4 that the decrease in the concentration in *Cu med.* is larger than that in *Ac* clouds by 1.75 times. The widest variation range of the ozone concentration is also in *Cu med.* The relative variation is the highest in *As* clouds—it reaches 27%. The minimum one (17%) has been fixed in *Cu*, *St*, and *Ac* clouds; this is seemingly caused by their water content.

### 3. ESTIMATE OF THE SCALE OF THE CLOUD EFFECT ON THE OZONE BALANCE IN THE TROPOSPHERE

The data given above regarding the variations of the ozone concentration in clouds allow for an estimation of the scale of ozone runoff from the troposphere when interacting with cloud aerosol in the first approximation.

According to [24], the average annual amount of clouds on the Earth is 5.8, i.e., 58% of the surface is covered by clouds. The area of the Earth is  $5.1 \times 10^{14}$  m<sup>2</sup>. Again, multiplying the thickness of the cloud cover (it equals 1.5 km in our experiments) by the area, we obtain a volume of  $4.44 \times 10^{17}$  m<sup>3</sup> where ozone runoff occurs.

According to Table 2, the variation of the ozone concentration is 11 ppb by the "minimum" and 15 ppb by the "maximum" ozonometers, or 22 and 30 µg/m<sup>3</sup>. Multiplying the volume by the concentration, we obtain 97.6 and 133.1 Tg of ozone precipitating on aerosol particles in clouds. The thus obtained estimate is close to the values of the annual ozone balance in the troposphere given in [25, 26] equal to 141 and 241 Tg per year, respectively (Table 5).

### CONCLUSIONS

Ozone destruction on cloud aerosol has been revealed from the results of the experiments carried out with the help of three ozonometers mounted onboard the "Optik-E" An-30 aircraft laboratory. The following types of clouds were studied: *Cu*, *Cu med.*, *St*, *Sc*, *As*, and *Ac*. The thickness of the cloud layers was 1.5 km on average and varied from 0.4 to 4.5 km.

Ozone destruction in clouds was 11–15 ppb on average and ranged from 3 to 34 ppb; it varied nearly twofold depending on the cloud type.

The estimation of the annual runoff of ozone in clouds has shown that it is close to the annual ozone balance in the troposphere.

### ACKNOWLEDGMENTS

This work was supported by the Presidium of the Russian Academy of Sciences (program no. 16), the Russian Foundation for Basic Research (project nos. 07-05-00645, 08-05-92499, and 09-05-10020), the International Science and Technology Center (project no. 3032), and the Federal Agency for Science and Innovations (state contract no. 05.218.11.7153).

### REFERENCES

1. S. P. Perov and A. Kh. Khrgian, *Modern Problems of Atmospheric Ozone* (Gidrometeoizdat, Leningrad, 1980) [in Russian].
2. S. V. Razumovskii and G. E. Zaikov, *Ozone and its Reactions with Organic Compounds* (Nauka, Moscow, 1974) [in Russian].
3. "Atmosphere Trace Gases that are Radiatively Active and Significance to Global Change," *Earth Quest.* **40** (2), 10–11 (1990).
4. B. D. Belan, "Tropospheric Ozone. 3. Ozone Abundance in Troposphere. Mechanisms and Factors," *Opt. Atmosf. Okeana* **21** (7), 600 (2008).
5. B. D. Belan and T. K. Sklyadneva, "Tropospheric Ozone. 4. Photochemical Formation of "Tropospheric Ozone: Role of Solar Radiation," *Opt. Atmosf. Okeana* **21**, 858 (2008).
6. B. D. Belan, "Tropospheric Ozone. 5. Gases—Precursors of Ozone," *Opt. Atmosf. Okeana* **22**, 230–268 (2009).
7. B. D. Belan, "Tropospheric Ozone. 6. Components of Main Cycles," *Opt. Atmosf. Okeana* **22**, 358–379 (2009).
8. J. Lelieveld and P. J. Crutzen, "The Role of Clouds in Tropospheric Photochemistry," *J. Atmos. Chem.* **12**, 229–267 (1991).
9. J. E. Jonson and I. S. A. Isaksen, "Tropospheric Ozone Chemistry. The Impact of Cloud Chemistry," *J. Atmos. Chem.* **16**, 99–122 (1993).
10. P. Brimblecombe, *Air Composition and Chemistry* (Cambridge Univ., 1986; Mir, Moscow, 1988).
11. J. Gonw and E. Lovejoy, "Reactive Uptake of Ozone by Liquid Organic Compounds," *Geophys. Res. Lett.* **25**, 931–934 (1998).
12. R. Sander, "Modeling Atmospheric Chemistry: Interactions between Gas-Phase Species and Liquid Cloud / Aerosol Particles," *Surv. Geophys.* **20**, 1–31 (1999).
13. Yu. M. Gershenzon, A. N. Ermakov, and A. P. Pural', "Chemical Reactions of Free Radicals with Atmospheric Aerosols," *Khim. Fiz.* **19** (3), 3 (2000).
14. G. Khaiklin, "Excretion of Gases from Atmosphere by Aerosol Particles," in *Heterogeneous Chemistry of Atmosphere* (Gidrometeoizdat, Leningrad, 1986), p. 171–182 [in Russian].
15. Z. Wang and K. Sassen, "Ozone Destruction in Continental Stratus Clouds: An Aircraft Case Study," *J. Appl. Meteorol.* **39**, 875–886 (2000).
16. D. Jacob, "Heterogeneous Chemistry and Tropospheric Ozone," *Atmos. Environ.* **34**, 2131–2159 (2000).
17. J. Reichardt, A. Ansmann, M. Serwazi, C. Weitkamp, and W. Michaelis, "Unexpectedly Low Ozone Concentration in Midlatitude Tropospheric Ice Clouds: A Case Study," *Geophys. Res. Lett.* **23**, 1929–1932 (1996).
18. M. Yu. Arshinov, B. D. Belan, O. A. Krasnov, V. K. Kovalenskii, V. A. Pirogov, A. P. Plotnikov, G. N. Tolmachev, and A. V. Fofonov, "Comparison of Ultraviolet and Chemiluminescent Ozone Analyzers," *Opt. Atm. Okeana* **15**, 723–726 (2002).
19. D. Grosjean and J. Harrison, "Response of Chemiluminescent NO<sub>x</sub> Analyzers and Ultraviolet Ozone Analyzers to Organic Air Pollutants," *Environ. Sci. and Technol.* **19** (9), 862–872 (1985).
20. E. E. Hudgens, T. E. Kleindienst, F. F. McElroy, and W. M. Ollison, "A Study of Interferences in Ozone UV and Chemiluminescence Monitors Measurement of Toxic and Related Air Pollutants," *J. Air Waste Manage. Assoc.* **44**, 405 (1994).

21. T. E. Kleindienst, E. E. Hudgens, and D. F. Smith, "Comparison of Chemiluminescence and Ultraviolet Ozone Monitor Responses in the Presence of Humidity and Photochemical Pollutants," *J. Air Waste Manage. Assoc.* **43**, 213 (1993).
22. J. J. Huntzicker and R. L. Johnson, "Investigation of An Ambient Interference in the Measurement of Ozone by Ultraviolet Absorption Photometry," *Environ. Sci. Technol.* **13**, 1414 (1979).
23. E. J. Dunlea, S. C. Herndon, D. D. Nelson, R. M. Volkamer, B. K. Lamb, E. J. Allwine, M. Grutter, C. R. Ramos Villegas, C. Marquez, S. Blanco, B. Cardenas, C. E. Kolb, L. T. Molina, and M. J. Molina, "Technical Note: Evaluation of Standard Ultraviolet Absorption Ozone Monitors in a Polluted Urban Environment," *Atmos. Chem. Phys.* **6**, 3163 (2006).
24. *Manual on Atmosphere* (Gidrometeoizdat, Leningrad, 1991) [in Russian].
25. P. G. Pruchniewich and P. Fabian, "Meridional Distributions of Ozone in Troposphere and its Seasonal Variations," *J. Geophys. Res. D* **82**, 2063 (1977).
26. F. Routhier, "Free Tropospheric and Boundary Layer Airborne Measurements of Ozone over the Latitude Range of 58° S to 70° N," *J. Geophys. Res. C* **85**, 2848–2960 (1980).