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Olga Yu. Antokhina, Pavel N. Antokhin, Mikhail Yu. Arshinov, Victoria G. Arshinova, Boris D.

Belan, Sergey B. Belan, Denis K. Davydov, Nina V. Dudorova, Georgii A. Ivlev, Artem V. Kozlov,

Olga V. Praslova*, Tatyana M. Rasskazchikova, Denis E. Savkin, Denis V. Simonenkov, Tatyana K.

Sklyadneva, Gennadii N. Tolmachev, Aleksander V. Fofonov

V.E. Zuev Institute of Atmospheric Optics of Siberian Branch of the Russian Academy of Sciences, 1, Academician Zuev sq., Tomsk, Russia, 634055

ABSTRACT

The paper studies the changes in the gas composition of the atmosphere in the surface layer for the Tomsk region (Akademgorodok) during the periods of sharp change of air masses. Studies were conducted for the summer seasons (June-August 1993-2018) for the periods of change of air masses from warm to cold. Ozone, carbon dioxide, carbon monoxide and sulfur dioxide (O₃, CO₂, CO, SO₂) concentrations were used for the analysis of the change in chemical composition. The concentrations have been derived from TOR station. For the analysis of air mass change, ECMWF Era-Interim data on the potential temperature on the dynamic tropopause (PV- θ) was used. The paper studies the processes air mass change in which the difference of PV- θ anomaly was about 20-30 K. It was found that the concentration of CO₂ is mainly increasing, and concentrations of O₃ and aerosol are decreasing during the polar intrusion. For CO and SO₂, no dependence on a sharp change in the air mass properties was found.

Keywords: atmospheric composition, gas, aerosol, air masses, polar intrusion, PV-0.

1. INTRODUCTION

A large number of works demonstrate heterogeneity in the chemical composition of air masses of different origin ^{1-3, 7, 21,} 28-32, 34, 35, 28-40. Areas of mid-latitudes from 50 to 70 ° N. experience the most drastic fluctuations in temperature, humidity and chemical composition due to a sharp change in air masses, especially when passing the fronts. The increase in horizontal geopotential gradients (>20 dkm/1000 km) on the axis of the dynamically significant high-altitude frontal zone at the counter advection of cold and heat (\geq 4-5 °C/day) to the rear and front of the cyclone, respectively, should be considered as signs of sharp changes and the formation of extreme weather phenomena. In a large number of works, the influence of a sharp change in air masses and the passage of fronts was studied ^{6,9,10,12,23,26,27,36,37}. In some studies ^{1,2,16} investigated the effect of the passage of fronts on the concentration of ozone and aerosol, it was shown that the atmospheric fronts significantly affect the concentration of ozone and aerosol. Depending on the geographical type of the front and the direction of its movement patterns of ozone distribution are modified. A number of works are devoted to the processes of blocking ^{11,13}, anomalies of concentrations of small gas components ^{15, 22}, modeling and forecasting of atmospheric processes ^{14, 41, 42}. Also, the influence of synoptic conditions on the aerosol optical thickness is investigated ^{3, 24, 33}. The authors have shown that different air masses make it possible to distinguish different levels of turbidity, and, consequently, the change in the concentrations of gas components. The changes of CO_2 concentration in the territory of Western Siberia during the passage of atmospheric fronts in the warm and cold period of the year are estimated ⁴. The relationship between the distribution of carbon dioxide and the life of plants in the summer is shown. The following works show the influence of air mass transfer directions on the seasonal cycle of concentrations of small gas components of the atmosphere 7, 9, 16, 18-21, 23. In one of the works, according to the data of long-term monitoring at the TOR station, the average concentrations of gas and aerosol components in different air masses in the Tomsk region were calculated ⁵.

*pov@iao.ru; phone +7 3822 492-894; fax +7 3822 492-086; <u>http://lop.iao.ru/eng/</u>

25th International Symposium On Atmospheric and Ocean Optics: Atmospheric Physics, edited by Gennadii G. Matvienko, Oleg A. Romanovskii, Proc. of SPIE Vol. 11208, 1120885 © 2019 SPIE · CCC code: 0277-786X/19/\$21 · doi: 10.1117/12.2540920 It is shown that CO_2 and CH_4 are characterized by a decrease in the concentration from the Arctic to the tropical air mass. For ozone, the reverse is observed: the highest values are recorded in the tropical air mass, the lowest – in the Arctic. In gases such as CO and SO₂, the nature of the distribution is more complex.

In the present work, in contrast to previous studies, we set a more specific task. As the conditions, we will select those under which there is a sharp change of subtropical air, advection of which occurs from the crests of high pressure of large amplitude, from the territory of Central Asia (Mongolia, Kazakhstan, Western China) to the polar. Polar air masses come most often from the territory of the Kara and Barents seas and adjacent areas in systems of deep hollows. We will explore the summer period from June to August as the events described above are best explored for the individual seasons of the year. Our goal is to demonstrate the maximum changes that the gas and aerosol composition is capable of under the influence of changing air masses.

2. DATA AND TECHNIQUE

2.1 Chemical composition data set

Surface ozone (O3), carbon dioxide (CO2), carbon oxide (CO), sulfur dioxide (SO2) concentration, methane (CH4) and air temperature (T) for 1993-2018 have been derived from a TOR-station (Tropospheric Ozone Research). The TOR-station coordinates are 56°28'41"N., 85°03'15" E. This is an automatic station located in the Akademgorodok of Tomsk belongs to the V.E. Zuev Institute of Atmospheric Optics of Siberian Branch of the Russian Academy of Sciences (IAO SB RAS). There are no industrial facilities or motorways near the station, which reduces the influence of local sources of gas and aerosol. The station is located in the zone of boreal forests and surrounded by small woodland of deciduous and coniferous species. When air masses come from the west, the station is influenced by air that passes through Tomsk and contains industrial and transport pollutants. Otherwise, the air comes from areas with numerous forest stands and no large industrial enterprises. The technical equipment of TOR station demonstrates in Table 1. More detailed information about this station and the equipment can be found in [8].

| Block | Device/sensor | Measured parameter | Range | Error | Time constant, s | |
|--------------|-----------------------|---|----------|-------------|------------------|--|
| | | | | | | |
| Gas analyzer | LGR | CO ₂ , ppm | 2010000 | 0,2 ppm | | |
| block | Model 907-0010 | CH4, ppm | 0,00550 | 0,001 ppm | 1 | |
| | | H ₂ O, ppm | 15070000 | 100 ppm | | |
| | OPTEK 3.02 P-A | O ₃ , μ g/m ³ | 0500 | $\pm 20\%$ | 1 | |
| | Teledyne API 200E | NO/NO ₂ , ppm | 020 | $\pm 0,5\%$ | 20 | |
| | OPTEK K-100 | CO, mg/m ³ | 050 | $\pm 20\%$ | 1 | |
| | Teledyne API 100E | SO ₂ , ppm | 020 | $\pm 0,5\%$ | 20 | |
| | Brewer 049 | T.O.C, DU | - | $\pm 1\%$ | 120 | |
| Aerosol | GRIMM | D _p , micrometer (31 | | | | |
| Block | Model 1.109 | channel) | 0,2532 | - | | |
| | | N, cm ⁻³ | 02000 | ± 3% | 6 | |
| | Diffusive | D _p , nm (20 channels) | 3200 | - | | |
| | aerosol spectrometer | N, cm ⁻³ | 0500000 | $\pm 10\%$ | 80 | |

Table 1. Technical characteristics of the equipment TOR station.

2.2 Analyzing air masses

We used a new approach to the analysis of changes in air masses (respectively, advection of warm and cold air, the passage of fronts). For the analysis we used maps of the potential temperature at the dynamic tropopause (PV- θ) for 6 UTC. It is known that for a number of reasons [Masato] PV- θ is the best indicator of air masses possessing different properties. Low PV- θ values correspond to Arctic and polar air masses (dark blue in Fig. 1.), high values - tropical and subtropical (red in Fig. 1.).



Figure 1. Average monthly PV $-\theta$ (K) in June (a) and July (b) (1979-2018).

For analysis, we investigated PV- θ anomalies, for each day from June 1 to August 31, 1993-2018 relative to their average seasonal value in these months (separately June, July, August). We selected the events when the difference in the anomalies of PV- θ adjacent days was of the order of 20-30 Kelvin. Maps of anomalies in neighboring days and the date of the change is given in section 3 on the results. The chemical composition was considered three days before and two days after the date of change of air masses. Table 2 is composed for all dates. This table shows the gradients and daily rate of change for temperature, aerosol and gas concentrations. For example, CO₂ concentration mainly increases. It should be noted that abnormally high concentrations of CO₂ in 2010 are associated with a blocking anticyclone, which led to a large number of fires this year. The concentration of carbon monoxide can both increase and decrease. But the number of cases, when CO increases, slightly more than when decreases. The distribution of SO₂ is quite complex. In some cases, the peak concentration of SO₂ occurs at the time of change of air mass. The concentration of O₃ and aerosol decreases in all cases.

| dd/mm/y | ΔT °C | ℃/d | ΔΟ3, | V O ₃ | ∆a025 | V a025 | ΔCO_2 | V CO ₂ | ΔCO | V CO | ΔSO_2 | V SO ₂ |
|-------------------|----------|------|------|------------------|-------|--------|---------------|-------------------|--------|--------|---------------|-------------------|
| 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 |
| 26-28/07/ 1993 | 12 | -4 | 100 | -33.3 | - | - | - | - | 0.6412 | 0.2137 | - | - |
| 29-30/08/ 1993 | 13 | -6.5 | 44 | -22 | - | - | - | - | 1.9775 | 0.9887 | - | - |
| 01-02/07/ 1994 | 16 | -8 | 54 | -27 | - | - | - | - | - | - | - | - |
| 07-09/06/ 1994 | 17 | -5.5 | 42 | -14 | - | - | - | - | - | - | - | - |
| 14-16/08/ 1994 | 12 | -4 | 41 | -13.6 | - | - | - | - | - | - | - | - |
| 10-12/06/ 1996 | 12 | -4 | 60 | -20 | - | - | - | - | - | - | - | - |
| 23-26/06/ 1996 | 11 | -5.5 | 80 | -40 | - | - | - | - | - | - | - | - |
| 08-09/06/ 1997 | 18 | -9 | - | - | - | - | - | - | 0.0206 | 0.0103 | - | - |
| 29-30/06/ 1997 | 11 | -5.5 | 106 | -53 | - | - | - | - | - | - | - | - |
| 05-07/07/ 1997 | 13 | -4.3 | 60 | -20 | - | - | - | - | - | - | - | - |
| 16-19/07/ 1997 | 14 | -4.6 | 69 | -23 | - | - | - | - | - | - | - | |

Table 2. Gradients and daily rate of changes in T; aerosol, O3, CO2, CO, SO2 concentrations.

| 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 |
|--------------------------|----|-----------|-----|-------|---|---|---|---|--------|---------|----|----|
| 08-10/08/ 1997 | 19 | -6.3 | 95 | -31.6 | - | - | - | - | - | - | - | - |
| 17-20/08/ 1997 | 14 | -3.5 | 18 | -4.5 | - | - | - | - | - | - | - | - |
| 02-03/06/ 1998 | 18 | -9 | 37 | -18.5 | - | - | - | - | 0.115 | -0.0575 | - | - |
| 08-10/06/ 1998 | 11 | -3.6 | 23 | -7.6 | - | - | - | - | 0.0055 | 0.0018 | - | - |
| 16-17/08/ 1998 | 14 | -7 | 12 | -6 | - | - | - | - | 0.1033 | -0.516 | - | - |
| 08-10/06/ 1999 | 19 | - 6.33 | 28 | -9.3 | - | - | - | - | 0.0387 | -0.0129 | - | - |
| 15-17/06/ 1999 | 12 | -4 | 39 | -13 | - | - | - | - | 0.0826 | -0.0275 | - | - |
| 07-08/07/ 2000 | 14 | -7 | 64 | -32 | - | - | - | - | - | - | - | - |
| 01-02/06/ 2001 | 12 | -6 | 126 | -63 | - | - | - | - | 0.0029 | -0.0014 | - | - |
| 28/06- 01/07/2001 | 15 | -3.7 | 54 | -13.6 | - | - | - | - | - | - | - | - |
| 18-21/08/ 2001 | 10 | -2.5 | 178 | -44.5 | - | - | - | - | 0.0327 | -0.0081 | - | - |
| 11-12/08/ 2002 | 14 | -7 | 94 | -47 | - | - | - | - | 0.0254 | -0.0127 | - | - |
| 20-23/08/ 2002 | 14 | -3.5 | 29 | -7.2 | - | - | - | - | 0.0618 | -0.0154 | - | - |
| 07-08/06/ 2003 | 13 | -6.5 | 39 | -19.5 | - | - | - | - | 0.0123 | -0.0061 | - | - |
| 16-18/06/ 2003 | 10 | -3.3 | 9 | -3 | - | - | - | - | 0.0469 | 0.0156 | - | - |
| 08-11/07/ 2003 | 18 | -4.5 | 100 | -25 | - | - | - | - | 0.0952 | -0.0238 | - | - |
| 22-23/07/ 2003 | 12 | -6 | 61 | -30.5 | - | - | - | - | 0.0016 | -0.0008 | - | - |
| 31/07- 02/08/ 2003 | 12 | -4 | 35 | -11.6 | - | - | - | - | 0.0637 | 0.0159 | - | - |
| 05-08/06/ 2004 | 14 | -3.5 | 37 | -9.2 | - | - | - | - | 0.0085 | -0.0021 | - | - |
| 22-24/08/ 2005 | 17 | -5.6 | 43 | -14.3 | - | - | - | - | 0.0103 | 0.0034 | - | - |
| 12-14/06/ 2006 | 14 | -4.6 | 37 | -12.3 | - | - | - | - | 0.099 | 0.033 | - | - |
| 04-06/07/ 2006 | 17 | -5.6 | 45 | -15 | - | - | - | - | 0.0215 | -0.0071 | - | - |
| 23-24/07/ 2006 | 13 | -6.5 | 48 | -24 | - | - | - | - | 0.0044 | -0.0022 | - | - |
| 14-15/08/ 2006 | 10 | -5 | 66 | -33 | - | - | - | - | 0.0039 | -0.0019 | - | - |
| 24-26/06/ 2007 | 14 | -4,6 | 46 | -15.3 | - | - | - | - | 0.0056 | 0.0018 | - | - |
| 03-06/06/ 2008 | 13 | -3.2 | 56 | -14 | - | - | - | - | 0.0441 | -0.011 | - | - |
| 22-24/06/ 2008 | 10 | -2.3 | 35 | -11.6 | - | - | - | - | 0.0325 | 0.2006 | - | - |
| 22-23/07/ 2008 | 10 | -5 | 8 | -4 | - | - | - | - | 0.053 | 0.0265 | - | - |

| 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 |
|--------------------------|----|------|-----|--------|--------|--------------|------------|--------|--------|---------|---------|-------------|
| 10-11/08/ 2008 | 11 | -5.6 | 32 | -16 | - | - | - | - | 0.0615 | -0.0307 | - | - |
| 15-18/08/ 2008 | 12 | -3 | 62 | -15.5 | - | - | - | - | 0.0933 | 0.233 | - | - |
| 10-13/06/ 2009 | 16 | -4 | 88 | -22 | - | - | - | - | 0.0188 | -0.0047 | - | - |
| 11-12/06/ 2010 | 12 | -6 | 35 | -11.6 | 90307 | 30102.5 | 5.754 | 2.877 | 0.0872 | 0.0436 | - | - |
| 20-22/06/ 2010 | 21 | -7 | 91 | -30.3 | 66268 | 22089.5 | 15.06 2 | 5.020 | 0.0766 | 0.0255 | - | - |
| 17-20/07/ 2010 | 12 | -3 | 110 | -27.5 | 163001 | 40750.3 | 22.98 5 | 7.661 | 0.0501 | -0.0125 | - | - |
| 22-24/08/ 2010 | 13 | -4.3 | 56 | -18.6 | - | - | 13.73 7 | 4.579 | 0.0422 | 0.0140 | - | - |
| 01-02/07/ 2011 | 12 | -6 | 57 | -28.5 | 189692 | 94846 | - | - | 0.0229 | 0.0114 | 0.8943 | -0.4471 |
| 03-05/08/ 2011 | 12 | -4 | 41 | -13.6 | 125557 | 41852.3 | 5.813 | 1.937 | 0.016 | -0.0053 | 10.7569 | -3.5856 |
| 31/07- 03/08/ 2012 | 13 | 3.25 | 58 | -19.3 | 327641 | 81910.2 | - | - | 1.1389 | 0.2847 | - | - |
| 15-16/06/ 2013 | 15 | -7.5 | 21 | -10.5 | 74983 | 37491.6 | - | - | 0.1111 | 0.0111 | 0.0555 | 0.1315 |
| 07-08/07/ 2013 | 12 | -6 | 30 | -15 | 221943 | 110971. 8 | 0.638 | -0.319 | 0.0889 | 0.0444 | 0.9706 | -0.4853 |
| 30-31/08/ 2013 | 12 | -6 | 29 | -14,5 | 90418 | 30139.3 | 1.293 | -0.646 | 0.0843 | 0.0421 | 2.2961 | 1.1480 |
| 25-27/08/ 2014 | 17 | -5.6 | 56 | -18.6 | 88316 | 29438.7 | - | - | 0.0175 | 0.058 | 1.1746 | -0.3915 |
| 7-10 /06/ 2015 | 16 | -4 | 18 | -4.5 | 110290 | 27572.6 | 4.518 | 1.129 | 0.2862 | 0.0715 | 1.404 | ±0.351 |
| 03-05/07/ 2015 | 21 | -4.2 | 43 | -14.3 | 274345 | 91448 | - | - | 0.2385 | -0.0795 | 2.9954 | ±0.998 4 |
| 19-22/08/ 2015 | 18 | -4.5 | 53 | -13.25 | 25805 | 6451.3 | 6.778 | 1.694 | 0.0918 | 0.0226 | 2.8989 | ±0.724 7 |
| 05-08/08/ 2017 | 12 | -3 | 44 | -11 | 136943 | 34235.7 | - | - | 0.0261 | 0.0652 | 11.6154 | -2.9038 |
| 2- 4/07/2018 | 14 | -4.6 | 43 | -14.3 | 74734 | 24911.3 | - | - | 0.0346 | 0.0115 | 0.5463 | ±0.182 1 |

3. RESULTS

As a result of this work, on the basis of TOR station data, changes in the concentrations of O_3 , CO_2 , CO, SO_2 and aerosol during the passage of polar air masses were investigated. The dates presented in Table 2 are selected as the events of a sharp change in air masses. A certain correlation of the concentration of some gases from the temperature was revealed. The higher the rate of daily temperature change, the higher the rate of change in O_3 concentration, while the concentration decreases. For CO_2 , the situation is less straightforward, as the summer CO_2 concentration is influenced by vegetation activity and fire conditions. However, it should be noted that carbon dioxide is characterized by a reverse gradient with respect to ozone. For carbon monoxide, as well as for sulfur dioxide, the temperature dependence could not be detected. In 25 cases, out of 46, the concentration of CO increases, and in 21 cases – decreases with the change of air mass. From a number of observations (11 cases) for SO_2 , 5 cases were revealed when its concentration decreases, 2 cases when it increases and 4 cases when the concentration has an indefinite course. For a good example, in Figure 2, presents the cases of polar intrusion: June 8-10, 2015 July 3-5, 2015, July 2-4, 2018.

Proc. of SPIE Vol. 11208 1120885-5



Figure 2. Maps of anomalies of PV-θ (K) for 8-10 June 2015, 3-5 July 2015, 2-4 July 2018.

4. CONCLUSION

Based on the results obtained, it can be concluded that in the summer, polar intrusion have a significant impact not only on the change in the concentrations of aerosol and gas components, but also on the rate of its change. The distribution of SO_2 and CO requires further investigation. Also, sudden changes in temperature can, including cause the rate of transformation of the studied gases into other compounds, which can often have more aggressive chemical properties.

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